COURSE OUTLINE

(1) **GENERAL**

SCHOOL	School of Engineering			
ACADEMIC UNIT	Department of Naval Architecture			
LEVEL OF STUDIES	Undergraduate			
COURSE CODE	NAOME1217		SEMESTER	3 rd
COURSE TITLE	THERMODYNAMICS			
INDEPENDENT TEACHING ACTIVITIES		WEEKLY TEACHING HOURS	CREDITS (ECTS)	
Lectures		4	Л	
			4	
COURSE TYPE		General background		
general background,				
specialbackground, specialised general knowledge, skills development				
PREREQUISITE COURSES:				
LANGUAGE OF INSTRUCTION		Greek		
and EXAMINATIONS:				
IS THE COURSE OFFERED TO		Yes (Italian)		
ERASMUS STUDENTS				
COURSE WEBSITE (URL)		https://eclass.uniwa.gr/courses/NAFP111/		
		http://ocp.teiath.g	r/courses/NAFP_UNDER	110

(2) COURSE GOALS / LEARNING OUTCOMES

The aim of the course is to understand the basic principles and fundamentals of thermodynamics. During the course the students will be familiarized with the description and application of the physical concepts of work, heat, inner energy, temperature, entropy, the laws of thermodynamics and the use of tables and diagrams.

(3) COURSE CONTENT / SYLLABUS

1. Fundamental Concepts and Definitions : Terminology, definition and scope, microscopic and macroscopic approaches. Engineering Thermodynamics: Definition, some practical applications of engineering thermodynamics. System (closed system) and Control Volume (open system).

2. Ideal and real gases. Differences between ideal and real gases, equation of state for ideal gases, real gases. Van der Waal's equation of state, other equations of state.

3. The First Law of Thermodynamics. Basic concepts : system, state, equilibrium, process. Quasi – equilibrium processes. Equation of state.

4. Heat and Work: changing the state of a system. Zeroth law of thermodynamics. Work. The first law of thermodynamics and its corollaries: adiabatic, steady, throttling of a gas, quasi-static expansion of gas, transient filling of a tank. Enthalpy. Specific heats. Conservation of mass and energy in control volume form. Engineering cycles: properties of cycles, work and efficiency, general presentation of cycles, Carnot cycle, refrigerator and heat pump, Otto cycle, Diesel cycle, Joule cycle, Sabathe cycle.

5. The Second Law of Thermodynamics, Reversible processes, The second law of thermodynamics: statements and related concepts. Entropy changes in an ideal gas. Calculation of entropy change in basic processes.

6. Power cycles : Introduction, Practical Rankine Cycle, Reheat Cycle (continuation of

Rankine cycle), Regenerative Cycle. Mollier and Ts-diagrams.

(4) TEACHING and LEARNING METHODS - EVALUATION

DELIVERY Face-to-face, Distance learning, etc.	Face-to-face		
USE OF INFORMATION AND COMMUNICATIONS TECHNOLOGY Use of ICT in teaching, laboratory education, communication with students	 Support learning through class platform. 	the electronic e-	
TEACHING METHODS	Activity	Workload (hours)	
The manner and methods of teaching are	Lectures	39	
described in detail.	Exercises to understand	13	
fieldwork, study and analysis of	problems resolution		
bibliography, tutorials, placements, clinical	Personal study	65	
practice, art workshop, interactive teaching, educational visits project essay writing			
artistic creativity, etc.			
The student's study hours for each learning	Course total	117	
activity are given as well as the hours of non- directed study according to the			
principles of the ECTS			
STUDENT PERFORMANCE			
EVALUATION	Final written examination : 80%		
Description of the evaluation procedure	Evaluation of individual work : 2	20%	
evaluation, summative or conclusive, multiple			
choice questionnaires, short-answer questions,			
open-ended questions, problem solving, written			
presentation, laboratory work, clinical			
examination of patient, art interpretation,			
other			

(5) ATTACHED BIBLIOGRAPHY

- 1. Thermodynamic (theory and exercises), G Hatzikonstandis, UNIWA 2019
- 2. TERMODINAMICA E TRASMISSIONE DI CALORE, Y. Cangel & M. Boles, McGraw-Hill Education
- 3. Thermodynamics, ZEMANSKY, HOEPLI 2002
- 4. Fundamentals of Engineering Thermodynamics, MORAN & SHARPIRO, J. Wiley & Sons 2006